

References

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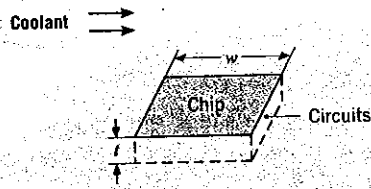
Problems

Conduction

- 1.1 The thermal conductivity of a sheet of rigid, extruded insulation is reported to be $k = 0.029 \text{ W/m} \cdot \text{K}$. The measured temperature difference across a 20-mm-thick sheet of the material is $T_1 - T_2 = 10^\circ\text{C}$.
 - (a) What is the heat flux through a $2 \text{ m} \times 2 \text{ m}$ sheet of the insulation?
 - (b) What is the rate of heat transfer through the sheet of insulation?
- 1.2 A concrete wall, which has a surface area of 20 m^2 and is 0.30 m thick, separates conditioned room air from ambient air. The temperature of the inner surface of the wall is maintained at 25°C , and the thermal conductivity of the concrete is $1 \text{ W/m} \cdot \text{K}$.
 - (a) Determine the heat loss through the wall for outer surface temperatures ranging from -15°C to 38°C , which correspond to winter and summer extremes, respectively. Display your results graphically.
 - (b) On your graph, also plot the heat loss as a function of the outer surface temperature for wall materials having thermal conductivities of 0.75 and $1.25 \text{ W/m} \cdot \text{K}$. Explain the family of curves you have obtained.
- 1.3 The concrete slab of a basement is 11 m long, 8 m wide, and 0.20 m thick. During the winter, temperatures are nominally 17°C and 10°C at the top and bottom surfaces, respectively. If the concrete has a thermal conductivity of $1.4 \text{ W/m} \cdot \text{K}$, what is the rate of heat loss through the slab? If the basement is heated by a gas furnace operating at an efficiency of $\eta_f = 0.90$ and natural gas is priced at $C_g = \$0.01/\text{MJ}$, what is the daily cost of the heat loss?
- 1.4 The heat flux through a wood slab 50 mm thick, whose inner and outer surface temperatures are 40 and 20°C , respectively, has been determined to be 40 W/m^2 . What is the thermal conductivity of the wood?
- 1.5 The inner and outer surface temperatures of a glass window 5 mm thick are 15 and 5°C . What is the heat loss through a window that is 1 m by 3 m on a side? The thermal conductivity of glass is $1.4 \text{ W/m} \cdot \text{K}$.
- 1.6 A glass window of width $W = 1 \text{ m}$ and height $H = 2 \text{ m}$ is 5 mm thick and has a thermal conductivity of $k_g = 1.4 \text{ W/m} \cdot \text{K}$. If the inner and outer surface temperatures of the glass are 15°C and -20°C , respectively, on a cold winter day, what is the rate of heat loss through the glass? To reduce heat loss through windows, it is customary to use a double pane construction in which adjoining panes are separated by an air space. If the spacing is 10 mm and the glass surfaces in contact with the air have temperatures of 10°C and -15°C , what is the rate of heat loss from a $1 \text{ m} \times 2 \text{ m}$ window? The thermal conductivity of air is $k_a = 0.024 \text{ W/m} \cdot \text{K}$.
- 1.7 A freezer compartment consists of a cubical cavity that is 2 m on a side. Assume the bottom to be perfectly insulated. What is the minimum thickness of styrofoam insulation ($k = 0.030 \text{ W/m} \cdot \text{K}$) that must be applied to the top and side walls to ensure a heat load of less than 500 W, when the inner and outer surfaces are -10 and 35°C ?
- 1.8 An inexpensive food and beverage container is fabricated from 25-mm-thick polystyrene ($k = 0.023 \text{ W/m} \cdot \text{K}$) and has interior dimensions of $0.8 \text{ m} \times 0.6 \text{ m} \times 0.6 \text{ m}$. Under conditions for which an inner surface temperature of approximately 2°C is maintained by an ice-water mixture and an outer surface temperature of 20°C is maintained by the ambient, what is the heat flux through the container wall? Assuming negligible heat gain through the $0.8 \text{ m} \times 0.6 \text{ m}$ base of the cooler, what is the total heat load for the prescribed conditions?
- 1.9 What is the thickness required of a masonry wall having thermal conductivity $0.75 \text{ W/m} \cdot \text{K}$ if the heat rate is to be 80% of the heat rate through a composite structural wall having a thermal conductivity of $0.25 \text{ W/m} \cdot \text{K}$ and a thickness of 100 mm? Both walls are subjected to the same surface temperature difference.
- 1.10 The 5-mm-thick bottom of a 200-mm-diameter pan may be made from aluminum ($k = 240 \text{ W/m} \cdot \text{K}$) or

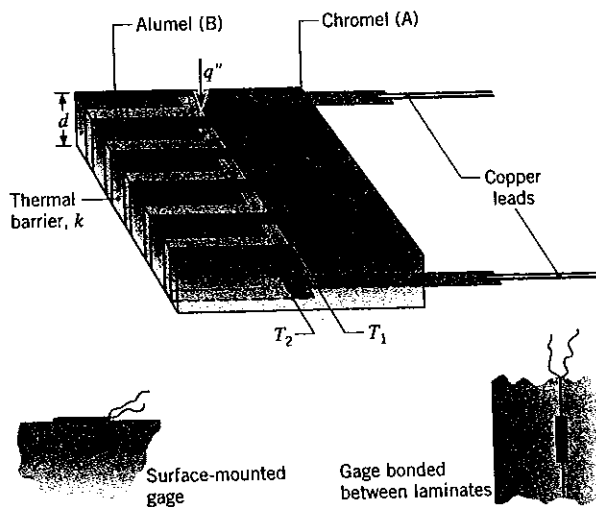
copper ($k = 390 \text{ W/m} \cdot \text{K}$). When used to boil water, the surface of the bottom exposed to the water is nominally at 110°C . If heat is transferred from the stove to the pan at a rate of 600 W , what is the temperature of the surface in contact with the stove for each of the two materials?

- 1.11 A square silicon chip ($k = 150 \text{ W/m} \cdot \text{K}$) is of width $w = 5 \text{ mm}$ on a side and of thickness $t = 1 \text{ mm}$. The chip is mounted in a substrate such that its side and back surfaces are insulated, while the front surface is exposed to a coolant.



If 4 W are being dissipated in circuits mounted to the back surface of the chip, what is the steady-state temperature difference between back and front surfaces?

- 1.12 A gage for measuring heat flux to a surface or through a laminated material employs five thin-film, chromel/alumel (type K) thermocouples deposited on the upper and lower surfaces of a wafer with a thermal conductivity of $1.4 \text{ W/m} \cdot \text{K}$ and a thickness of 0.25 mm .
- Determine the heat flux q'' through the gage when the voltage output at the copper leads is $350 \mu\text{V}$. The Seebeck coefficient of the type-K thermocouple materials is approximately $40 \mu\text{V}/^\circ\text{C}$.
 - What precaution should you take in using a gage of this nature to measure heat flow through the laminated structure shown?



Convection

- 1.13 You've experienced convection cooling if you've ever extended your hand out the window of a moving vehicle or into a flowing water stream. With the surface of your hand at a temperature of 30°C , determine the convection heat flux for (a) a vehicle speed of 35 km/h in air at -5°C with a convection coefficient of $40 \text{ W/m}^2 \cdot \text{K}$ and (b) a velocity of 0.2 m/s in a water stream at 10°C with a convection coefficient of $900 \text{ W/m}^2 \cdot \text{K}$. Which condition would *feel* colder? Contrast these results with a heat loss of approximately 30 W/m^2 under normal room conditions.

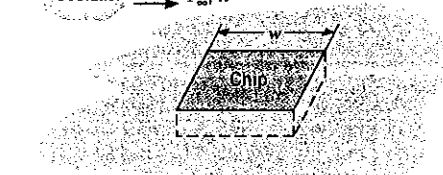
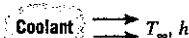
- 1.14 Air at 40°C flows over a long, 25-mm -diameter cylinder with an embedded electrical heater. In a series of tests, measurements were made of the power per unit length, P' , required to maintain the cylinder surface temperature at 300°C for different freestream velocities V of the air. The results are as follows:

Air velocity, V (m/s)	1	2	4	8	12
Power, P' (W/m)	450	658	983	1507	1963

- Determine the convection coefficient for each velocity, and display your results graphically.
 - Assuming the dependence of the convection coefficient on the velocity to be of the form $h = CV^n$, determine the parameters C and n from the results of part (a).
- 1.15 An electric resistance heater is embedded in a long cylinder of diameter 30 mm . When water with a temperature of 25°C and velocity of 1 m/s flows crosswise over the cylinder, the power per unit length required to maintain the surface at a uniform temperature of 90°C is 28 kW/m . When air, also at 25°C , but with a velocity of 10 m/s is flowing, the power per unit length required to maintain the same surface temperature is 400 W/m . Calculate and compare the convection coefficients for the flows of water and air.
- 1.16 A cartridge electrical heater is shaped as a cylinder of length $L = 200 \text{ mm}$ and outer diameter $D = 20 \text{ mm}$. Under normal operating conditions the heater dissipates 2 kW while submerged in a water flow that is at 20°C and provides a convection heat transfer coefficient of $h = 5000 \text{ W/m}^2 \cdot \text{K}$. Neglecting heat transfer from the ends of the heater, determine its surface temperature T_s . If the water flow is inadvertently terminated while the heater continues to operate, the heater surface is exposed to air that is also at 20°C but for which $h = 50 \text{ W/m}^2 \cdot \text{K}$. What is the corresponding surface temperature? What are the consequences of such an event?

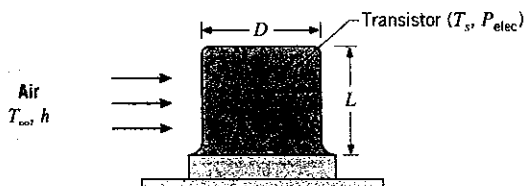
1.17 A common procedure for measuring the velocity of an air stream involves insertion of an electrically heated wire (called a *hot-wire anemometer*) into the air flow, with the axis of the wire oriented perpendicular to the flow direction. The electrical energy dissipated in the wire is assumed to be transferred to the air by forced convection. Hence, for a prescribed electrical power, the temperature of the wire depends on the convection coefficient, which, in turn, depends on the velocity of the air. Consider a wire of length $L = 20$ mm and diameter $D = 0.5$ mm, for which a calibration of the form, $V = 6.25 \times 10^{-5} h^2$, has been determined. The velocity V and the convection coefficient h have units of m/s and $W/m^2 \cdot K$, respectively. In an application involving air at a temperature of $T_\infty = 25^\circ C$, the surface temperature of the anemometer is maintained at $T_s = 75^\circ C$ with a voltage drop of 5 V and an electric current of 0.1 A. What is the velocity of the air?

1.18 A square isothermal chip is of width $w = 5$ mm on a side and is mounted in a substrate such that its side and back surfaces are well insulated, while the front surface is exposed to the flow of a coolant at $T_\infty = 15^\circ C$. From reliability considerations, the chip temperature must not exceed $T = 85^\circ C$.



If the coolant is air and the corresponding convection coefficient is $h = 200 W/m^2 \cdot K$, what is the maximum allowable chip power? If the coolant is a dielectric liquid for which $h = 3000 W/m^2 \cdot K$, what is the maximum allowable power?

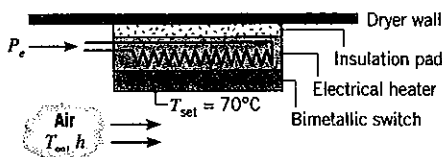
1.19 The case of a power transistor, which is of length $L = 10$ mm and diameter $D = 12$ mm, is cooled by an air stream of temperature $T_\infty = 25^\circ C$.



Under conditions for which the air maintains an average convection coefficient of $h = 100 W/m^2 \cdot K$ on the surface of the case, what is the maximum allowable power dissipation if the surface temperature is not to exceed $85^\circ C$?

1.20 The use of impinging air jets is proposed as a means of effectively cooling high-power logic chips in a computer. However, before the technique can be implemented, the convection coefficient associated with jet impingement on a chip surface must be known. Design an experiment that could be used to determine convection coefficients associated with air jet impingement on a chip measuring approximately 10 mm by 10 mm on a side.

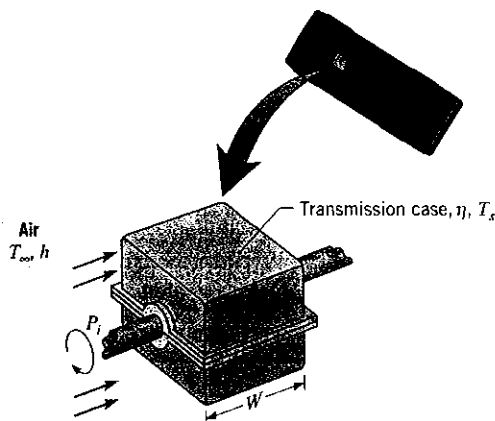
1.21 The temperature controller for a clothes dryer consists of a bimetallic switch mounted on an electrical heater attached to a wall-mounted insulation pad.



The switch is set to open at $70^\circ C$, the maximum dryer air temperature. In order to operate the dryer at a lower air temperature, sufficient power is supplied to the heater such that the switch reaches $70^\circ C$ (T_{set}) when the air temperature T is less than T_{set} . If the convection heat transfer coefficient between the air and the exposed switch surface of $30 mm^2$ is $25 W/m^2 \cdot K$, how much heater power P_e is required when the desired dryer air temperature is $T_\infty = 50^\circ C$?

1.22 The free convection heat transfer coefficient on a thin hot vertical plate suspended in still air can be determined from observations of the change in plate temperature with time as it cools. Assuming the plate is isothermal and radiation exchange with its surroundings is negligible, evaluate the convection coefficient at the instant of time when the plate temperature is $225^\circ C$ and the change in plate temperature with time (dT/dt) is $-0.022 K/s$. The ambient air temperature is $25^\circ C$ and the plate measures 0.3×0.3 m with a mass of 3.75 kg and a specific heat of $2770 J/kg \cdot K$.

1.23 A transmission case measures $W = 0.30$ m on a side and receives a power input of $P_i = 150$ hp from the engine.



Under conditions for which the air maintains an average convection coefficient of $h = 100 W/m^2 \cdot K$ on the surface of the case, what is the maximum allowable power dissipation if the surface temperature is not to exceed $85^\circ C$?

If the transmission efficiency is $\eta = 0.93$ and air flow over the case corresponds to $T_x = 30^\circ\text{C}$ and $h = 200 \text{ W/m}^2 \cdot \text{K}$, what is the surface temperature of the transmission?

1.29 If $T_s \approx T_{\text{sur}}$ in Equation 1.9, the radiation heat transfer coefficient may be approximated as

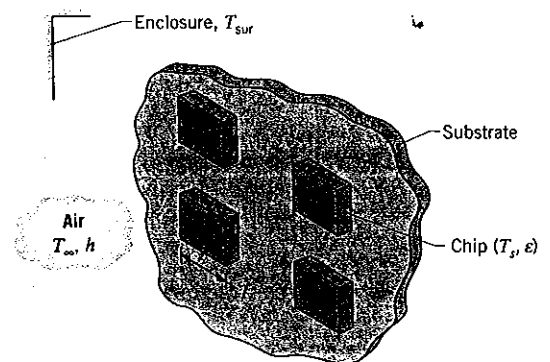
$$h_{r,a} = 4\epsilon\sigma\bar{T}^3$$

where $\bar{T} = (T_s + T_{\text{sur}})/2$. We wish to assess the validity of this approximation by comparing values of h_r and $h_{r,a}$ for the following conditions. In each case represent your results graphically and comment on the validity of the approximation.

Radiation

- 1.24** Under conditions for which the same room temperature is maintained by a heating or cooling system, it is not uncommon for a person to feel chilled in the winter but comfortable in the summer. Provide a plausible explanation for this situation (with supporting calculations) by considering a room whose air temperature is maintained at 20°C throughout the year, while the walls of the room are nominally at 27°C and 14°C in the summer and winter, respectively. The exposed surface of a person in the room may be assumed to be at a temperature of 32°C throughout the year and to have an emissivity of 0.90. The coefficient associated with heat transfer by natural convection between the person and the room air is approximately $2 \text{ W/m}^2 \cdot \text{K}$.
- 1.25** A spherical interplanetary probe of 0.5-m diameter contains electronics that dissipate 150 W. If the probe surface has an emissivity of 0.8 and the probe does not receive radiation from other surfaces, as, for example, from the sun, what is its surface temperature?
- 1.26** An instrumentation package has a spherical outer surface of diameter $D = 100 \text{ mm}$ and emissivity $\epsilon = 0.25$. The package is placed in a large space simulation chamber whose walls are maintained at 77 K. If operation of the electronic components is restricted to the temperature range $40 \leq T \leq 85^\circ\text{C}$, what is the range of acceptable power dissipation for the package? Display your results graphically, showing also the effect of variations in the emissivity by considering values of 0.20 and 0.30.
- 1.27** Consider the conditions of Problem 1.22. However, now the plate is in a vacuum with a surrounding temperature of 25°C . What is the emissivity of the plate? What is the rate at which radiation is emitted by the surface?
- 1.28** An overhead 25-m-long, uninsulated industrial steam pipe of 100 mm diameter is routed through a building whose walls and air are at 25°C . Pressurized steam maintains a pipe surface temperature of 150°C , and the coefficient associated with natural convection is $h = 10 \text{ W/m}^2 \cdot \text{K}$. The surface emissivity is $\epsilon = 0.8$.
- What is the rate of heat loss from the steam line?
 - If the steam is generated in a gas-fired boiler operating at an efficiency of $\eta_f = 0.90$ and natural gas is priced at $C_g = \$0.01$ per MJ, what is the annual cost of heat loss from the line?

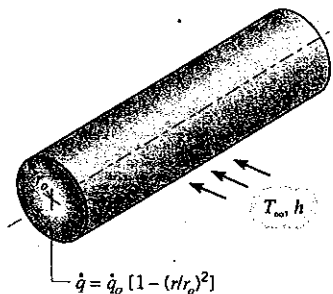
- Consider a surface of either polished aluminum ($\epsilon = 0.05$) or black paint ($\epsilon = 0.9$), whose temperature may exceed that of the surroundings ($T_{\text{sur}} = 25^\circ\text{C}$) by 10 to 100°C . Also compare your results with values of the coefficient associated with free convection in air ($T_\infty = T_{\text{sur}}$), where $h \text{ (W/m}^2 \cdot \text{K)} = 0.98 \Delta T^{1/3}$.
 - Consider initial conditions associated with placing a workpiece at $T_s = 25^\circ\text{C}$ in a large furnace whose wall temperature may be varied over the range $100 \leq T_{\text{sur}} \leq 1000^\circ\text{C}$. According to the surface finish or coating, its emissivity may assume values of 0.05, 0.2, and 0.9. For each emissivity, plot the relative error, $(h_r - h_{r,a})/h_r$, as a function of the furnace temperature.
- 1.30** Consider the conditions of Problem 1.18. With heat transfer by convection to air, the maximum allowable chip power is found to be 0.35 W. If consideration is also given to net heat transfer by radiation from the chip surface to large surroundings at 15°C , what is the percentage increase in the maximum allowable chip power afforded by this consideration? The chip surface has an emissivity of 0.9.
- 1.31** Chips of width $L = 15 \text{ mm}$ on a side are mounted to a substrate that is installed in an enclosure whose walls and air are maintained at a temperature of $T_{\text{sur}} = T_\infty = 25^\circ\text{C}$. The chips have an emissivity of $\epsilon = 0.60$ and a maximum allowable temperature of $T_s = 85^\circ\text{C}$.



- (a) If heat is rejected from the chips by radiation and natural convection, what is the maximum operating

- (a) For an initial condition corresponding to a wafer temperature of $T_{w,i} = 300$ K and the position of the wafer shown schematically, determine the corresponding time rate of change of the wafer temperature, $(dT_w/dt)_i$.
- (b) Determine the steady-state temperature reached by the wafer if it remains in this position. How significant is convection heat transfer for this situation? Sketch how you would expect the wafer temperature to vary as a function of vertical distance.

- 1.44** Radioactive wastes are packed in a long, thin-walled cylindrical container. The wastes generate thermal energy nonuniformly according to the relation $\dot{q} = \dot{q}_o[1 - (r/r_o)^2]$, where \dot{q} is the local rate of energy generation per unit volume, \dot{q}_o is a constant, and r_o is the radius of the container. Steady-state conditions are maintained by submerging the container in a liquid that is at T_∞ and provides a uniform convection coefficient h .

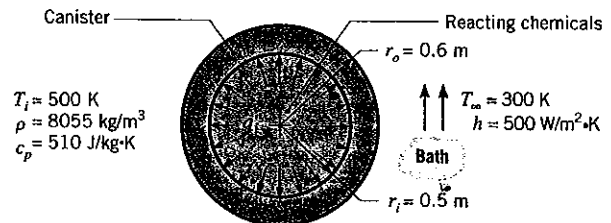


Obtain an expression for the total rate at which energy is generated in a unit length of the container. Use this result to obtain an expression for the temperature T_i of the container wall.

- 1.45** Consider the conducting rod of Example 1.3 under steady-state conditions. As suggested in Comment 3, the temperature of the rod may be controlled by varying the speed of air flow over the rod, which, in turn, alters the convection heat transfer coefficient. To consider the effect of the convection coefficient, generate plots of T versus I for values of $h = 50, 100,$ and 250 $\text{W/m}^2 \cdot \text{K}$. Would variations in the surface emissivity have a significant effect on the rod temperature?
- 1.46** A long bus bar (cylindrical rod used for making electrical connections) of diameter D is installed in a large conduit having a surface temperature of 30°C and in which the ambient air temperature is $T_\infty = 30^\circ\text{C}$. The electrical resistivity, ρ_e ($\mu\Omega \cdot \text{m}$), of the bar material is a function of temperature, $\rho_e = \rho_{e,o} [1 + \alpha(T - T_o)]$, where $\rho_{e,o} = 0.0171$ $\mu\Omega \cdot \text{m}$, $T_o = 25^\circ\text{C}$, and $\alpha = 0.00396$ K^{-1} . The bar experiences free convection in the ambient air, and the convection coefficient depends on the bar diameter, as well as on the difference between the surface and ambient

temperatures. The governing relation is of the form, $h = CD^{-0.25}(T - T_\infty)^{0.25}$, where $C = 1.21$ $\text{W} \cdot \text{m}^{-1.75} \cdot \text{K}^{-1.25}$. The emissivity of the bar surface is $\epsilon = 0.85$.

- (a) Recognizing that the electrical resistance per unit length of the bar is $R'_e = \rho_e/A_c$, where A_c is its cross-sectional area, calculate the current-carrying capacity of a 20-mm-diameter bus bar if its temperature is not to exceed 65°C . Compare the relative importance of heat transfer by free convection and radiation exchange.
- (b)** To assess the trade-off between current-carrying capacity, operating temperature, and bar diameter, for diameters of 10, 20, and 40 mm, plot the bar temperature T as a function of current for the range $100 \leq I \leq 5000$ A. Also plot the ratio of the heat transfer by convection to the total heat transfer.
- 1.47** A small sphere of reference-grade iron with a specific heat of 447 $\text{J/kg} \cdot \text{K}$ and a mass of 0.515 kg is suddenly immersed in a water-ice mixture. Fine thermocouple wires suspend the sphere, and the temperature is observed to change from 15 to 14°C in 6.35 s. The experiment is repeated with a metallic sphere of the same diameter, but of unknown composition with a mass of 1.263 kg. If the same observed temperature change occurs in 4.59 s, what is the specific heat of the unknown material?
- 1.48** A spherical, stainless steel (AISI 302) canister is used to store reacting chemicals that provide for a uniform heat flux q''_i to its inner surface. The canister is suddenly submerged in a liquid bath of temperature $T_\infty < T_i$, where T_i is the initial temperature of the canister wall.



- (a) Assuming negligible temperature gradients in the canister wall and a constant heat flux q''_i , develop an equation that governs the variation of the wall temperature with time during the transient process. What is the initial rate of change of the wall temperature if $q''_i = 10^5$ W/m^2 ?
- (b) What is the steady-state temperature of the wall?
- (c)** The convection coefficient depends on the velocity associated with fluid flow over the canister and whether or not the wall temperature is large enough to induce boiling in the liquid. Compute and plot the steady-state temperature as a function of h for

the range $100 \leq h \leq 10,000 \text{ W/m}^2 \cdot \text{K}$. Is there a value of h below which operation would be unacceptable?

1.49 Liquid oxygen, which has a boiling point of 90 K and a latent heat of vaporization of 214 kJ/kg, is stored in a spherical container whose outer surface is of 500-mm diameter and at a temperature of -10°C . The container is housed in a laboratory whose air and walls are at 25°C .

(a) If the surface emissivity is 0.20 and the heat transfer coefficient associated with free convection at the outer surface of the container is $10 \text{ W/m}^2 \cdot \text{K}$, what is the rate, in kg/s, at which oxygen vapor must be vented from the system?

(b) Moisture in the ambient air will result in frost formation on the container, causing the surface emissivity to increase. Assuming the surface temperature and convection coefficient to remain at -10°C and $10 \text{ W/m}^2 \cdot \text{K}$, respectively, compute the oxygen evaporation rate (kg/s) as a function of surface emissivity over the range $0.2 \leq \varepsilon \leq 0.94$.

1.50 A freezer compartment is covered with a 2-mm-thick layer of frost at the time it malfunctions. If the compartment is in ambient air at 20°C and a coefficient of $h = 2 \text{ W/m}^2 \cdot \text{K}$ characterizes heat transfer by natural convection from the exposed surface of the layer, estimate the time required to completely melt the frost. The frost may be assumed to have a mass density of 700 kg/m^3 and a latent heat of fusion of 334 kJ/kg.

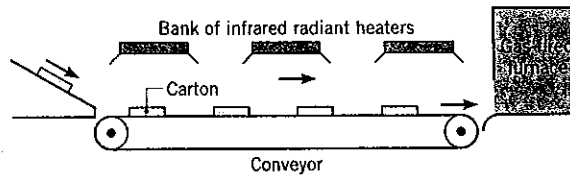
1.51 A vertical slab of Woods metal is joined to a substrate on one surface and is melted as it is uniformly irradiated by a laser source on the opposite surface. The metal is initially at its fusion temperature of $T_f = 72^\circ\text{C}$, and the melt runs off by gravity as soon as it is formed. The absorptivity of the metal to the laser radiation is $\alpha_l = 0.4$, and its latent heat of fusion is $h_{if} = 33 \text{ kJ/kg}$.

(a) Neglecting heat transfer from the irradiated surface by convection or radiation exchange with the surroundings, determine the instantaneous rate of melting in $\text{kg/s} \cdot \text{m}^2$ if the laser irradiation is 5 kW/m^2 . How much material is removed if irradiation is maintained for a period of 2 s?

(b) Allowing for convection to ambient air, with $T_\infty = 20^\circ\text{C}$ and $h = 15 \text{ W/m}^2 \cdot \text{K}$, and radiation exchange with large surroundings ($\varepsilon = 0.4$, $T_{\text{sur}} = 20^\circ\text{C}$), determine the instantaneous rate of melting during irradiation.

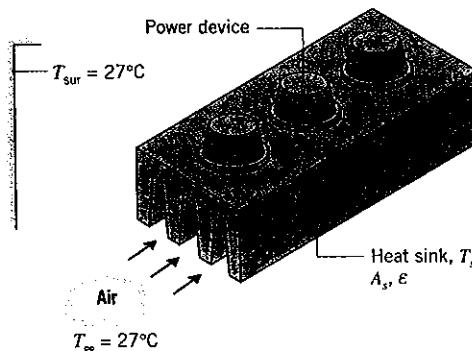
1.52 Following the hot vacuum forming of a paper-pulp mixture, the product, an egg carton, is transported on a conveyor for 18 s toward the entrance of a gas-fired

oven where it is dried to a desired final water content. It is observed that very little water evaporates during the travel time. So, to increase the productivity of the line, it is proposed that a bank of infrared radiation heaters, which provide a uniform radiant flux of 5000 W/m^2 , be installed over the conveyor. The carton has an exposed area of 0.0625 m^2 and a mass of 0.220 kg, 75% of which is water after the forming process.



The chief engineer of your plant will approve the purchase of the heaters if they can reduce the water content by 10% of the total mass. Would you recommend the purchase? Assume the heat of vaporization of water is $h_{fg} = 2400 \text{ kJ/kg}$.

1.53 Electronic power devices are mounted to a heat sink having an exposed surface area of 0.045 m^2 and an emissivity of 0.80. When the devices dissipate a total power of 20 W and the air and surroundings are at 27°C , the average sink temperature is 42°C . What average temperature will the heat sink reach when the devices dissipate 30 W for the same environmental condition?



1.54 A computer consists of an array of five printed circuit boards (PCBs), each dissipating $P_b = 20 \text{ W}$ of power. Cooling of the electronic components on a board is provided by the forced flow of air, equally distributed in passages formed by adjoining boards, and the convection coefficient associated with heat transfer from the components to the air is approximately $h = 200 \text{ W/m}^2 \cdot \text{K}$. Air enters the computer console at a temperature of $T_i = 20^\circ\text{C}$, and flow is driven by a fan whose power consumption is $P_f = 25 \text{ W}$.